

Lesson Title	Atomic Relationships: Radioactive Decay (2 class days)
Resources	Alpha Decay and Beta Decay labs, Crash Course Chemistry and Fermilab videos
Objectives	<ul style="list-style-type: none"> Distinguish among alpha, beta, gamma radiation types, and the relationship between the particles emitted by each, and the amount of energy emitted by each. Explain why radioactive decay happens. Describe the relationship between parent and daughter isotopes Describe the relationship between mass and energy.
Activities	<p>Day 1</p> <p>I do: Mini-talk (~30 min)</p> <ul style="list-style-type: none"> $E = mc^2$: What does this equation mean? (E and m are related. Atoms are tiny, so special relativity is totally cool here.) Show a uranium decay series. What's happening to mass through the series? Where does the mass go? Why do these atoms lose mass? Why do they need to lose energy? (Energetic stability) Videos from Crash Course Chemistry and Fermilab Define each type of decay, give examples, show nuclear equations. Point out parent vs daughter isotopes. Show cloud chamber pictures, describe how it works. <p>We do: Cloud chamber! (25 min)</p> <ul style="list-style-type: none"> I'll tell them which patterns correspond to which type of decay Take five minutes to make observations in notebook. No guiding questions here---anything goes: questions, pictures, descriptions, etc. What questions do you have? Informal group discussion for ~10 minutes. Debrief: Patterns are formed by radioactive decay from isotopes around us--such as natural radon (alpha) potassium in our bods and the earth (beta) cosmic rays (gamma)! <p>Regroup (10 min)</p> <ul style="list-style-type: none"> Introduce lab, have students partner up, distribute lab for them to take a look over. Exit ticket <ul style="list-style-type: none"> Name one type of radioactive decay. What type of particle is emitted? Is it low, middle, or high-energy? <p>Day 2</p> <p>Bellringer: Radium-226 undergoes alpha decay. What type of particle is emitted? What is the mass of the daughter isotope?</p> <p>I do: Recap and reintroduce lab (~15 minutes)</p> <ul style="list-style-type: none"> Go over bellringer Go back over tier-3 words Go back over lab instructions, take questions. <p>You do: Alpha and Beta Decay Lab (45 minutes)</p>

	<ul style="list-style-type: none"> • Students will work in small pairs to start their guided inquiry with the Alpha Decay and Beta Decay PhET simulations. • Guides are based on Trish Loblein's activities for Alpha Decay and Beta Decay.
Assessment	Exit tickets/bellringers, labs
Differentiation	Advance organizers for note-taking and vocab. Accept one-word and diagrammed answers for ELL. Lecture is long, so maybe take a couple movement breaks. If extra time on lab needed, can come in and work on it after class/during lunch/after school/etc.